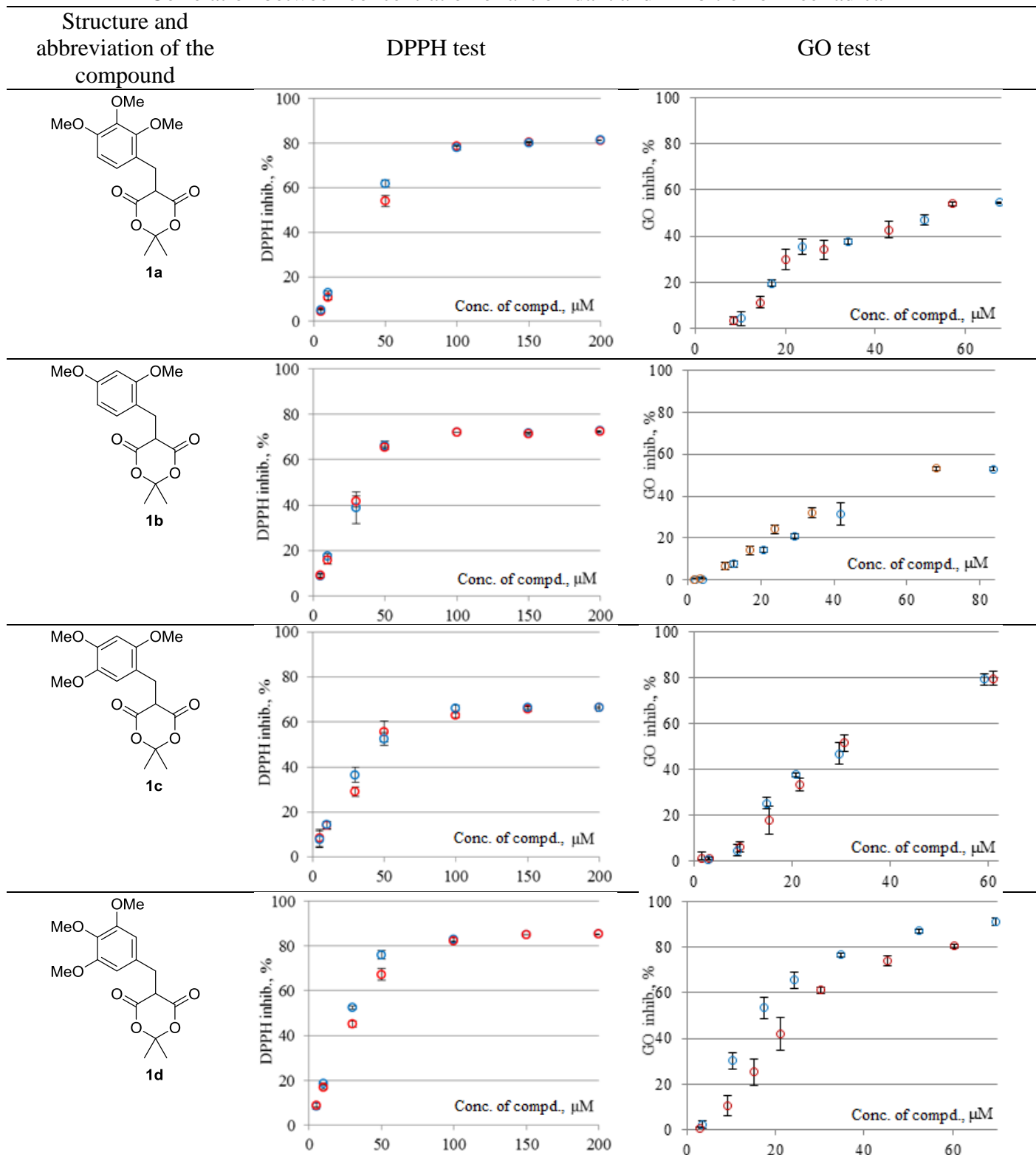
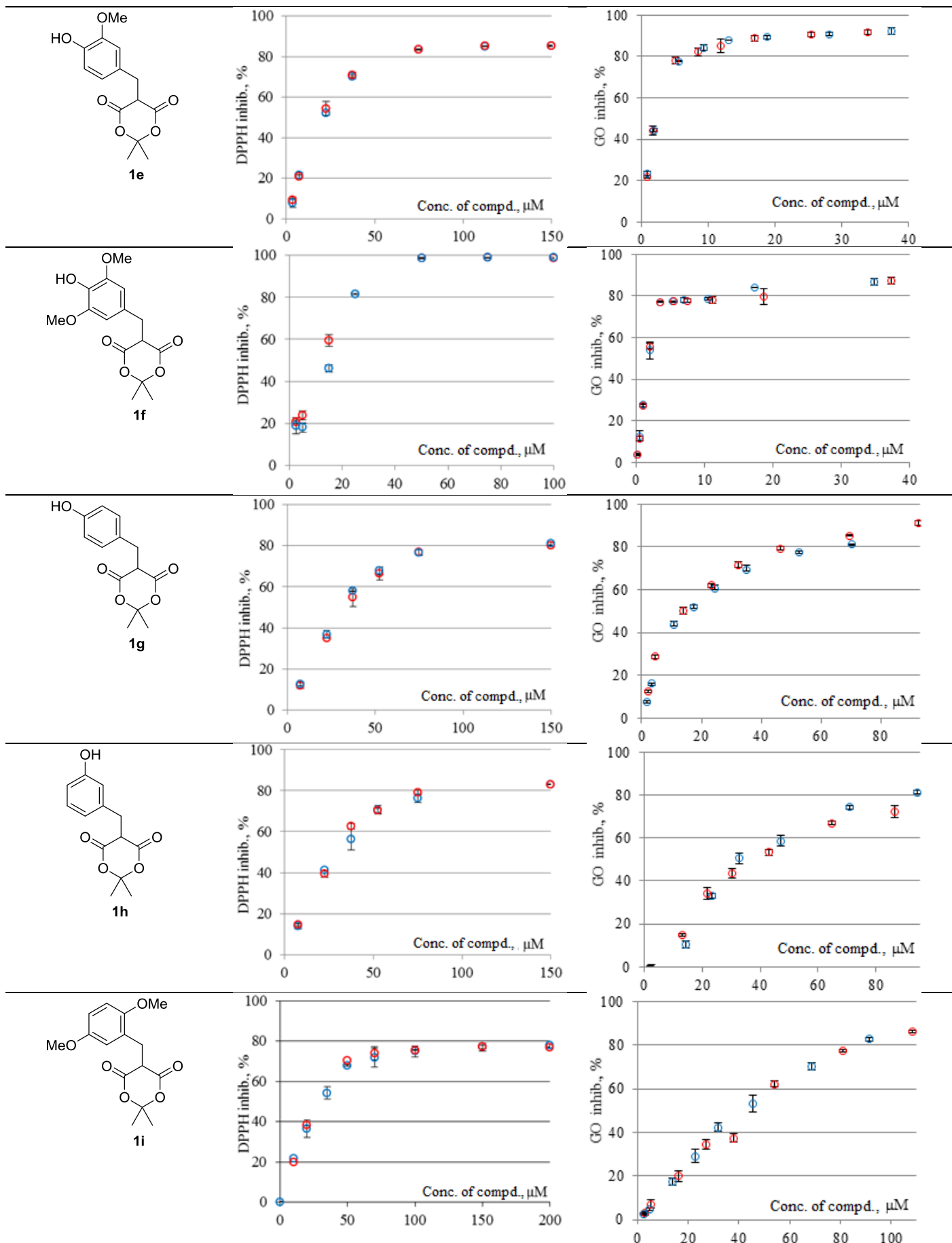
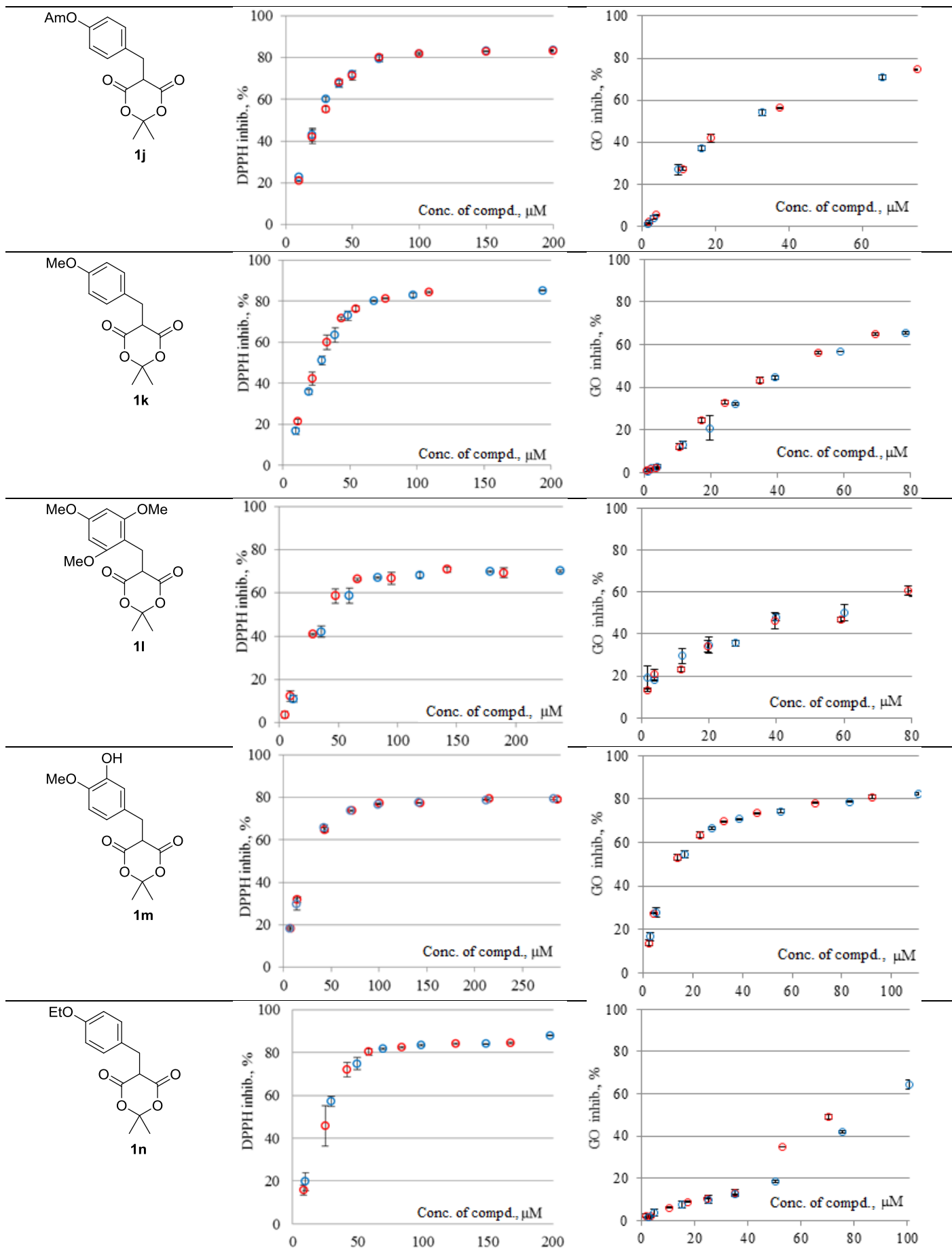


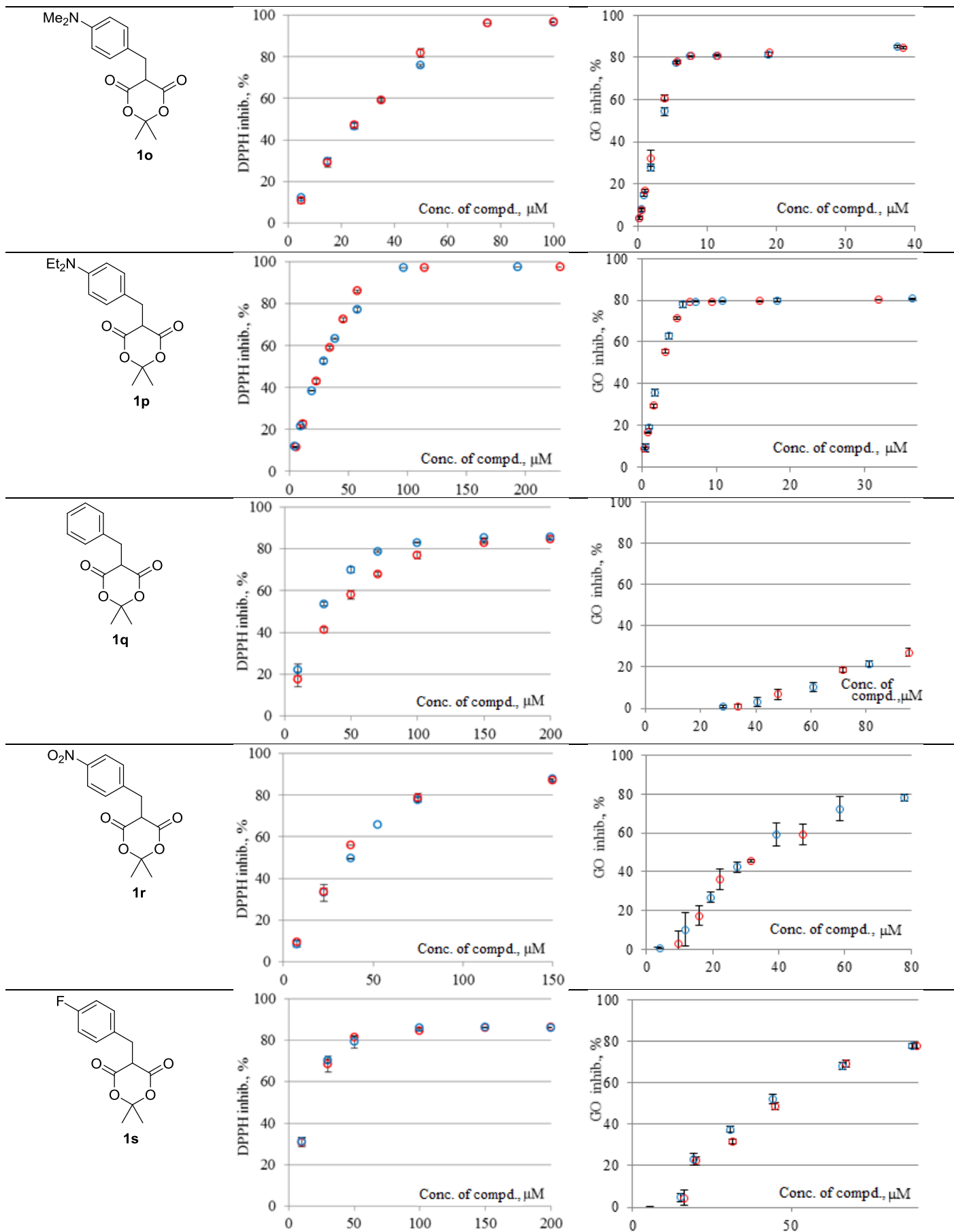
Correlation between concentration of antioxidant and inhibition of free radical*

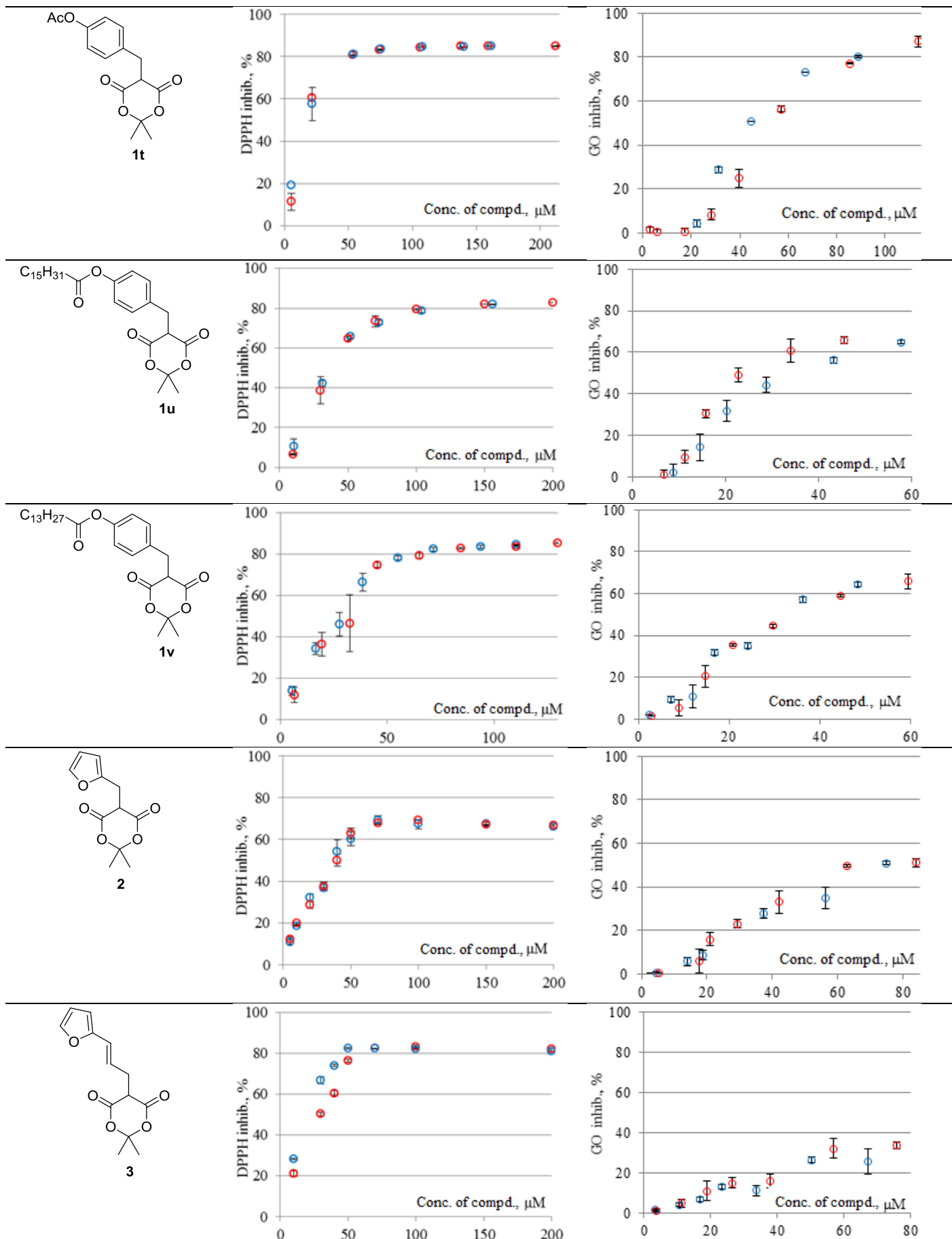


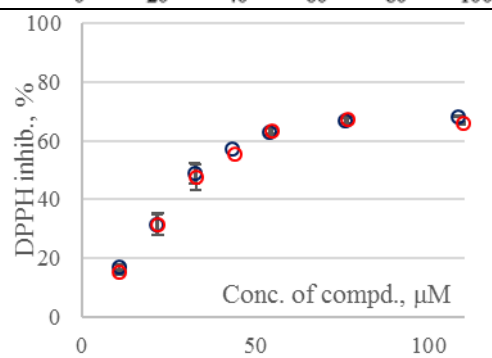
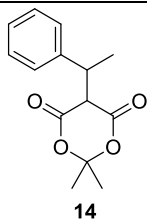
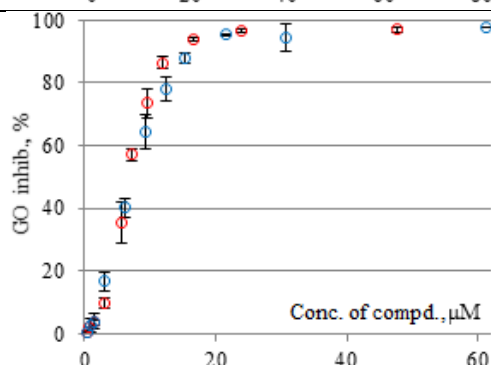
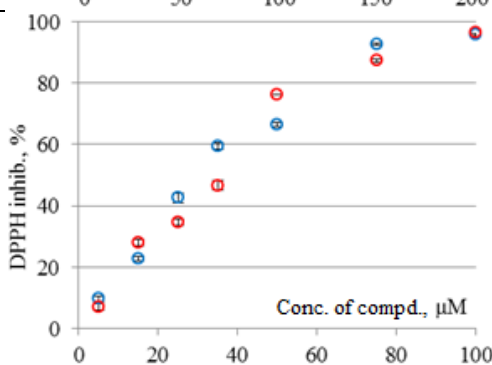
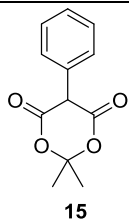
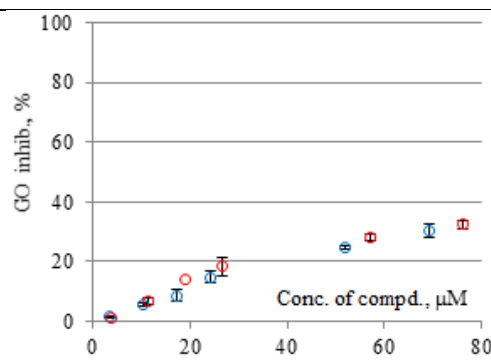
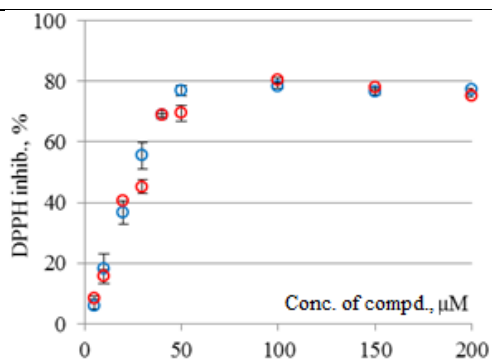
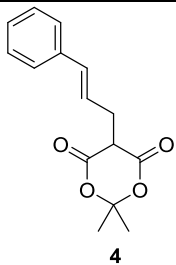
* The data are presented as a mean \pm standard deviation from three measurements. Different colors of bulletins present results from two unattached experiments.











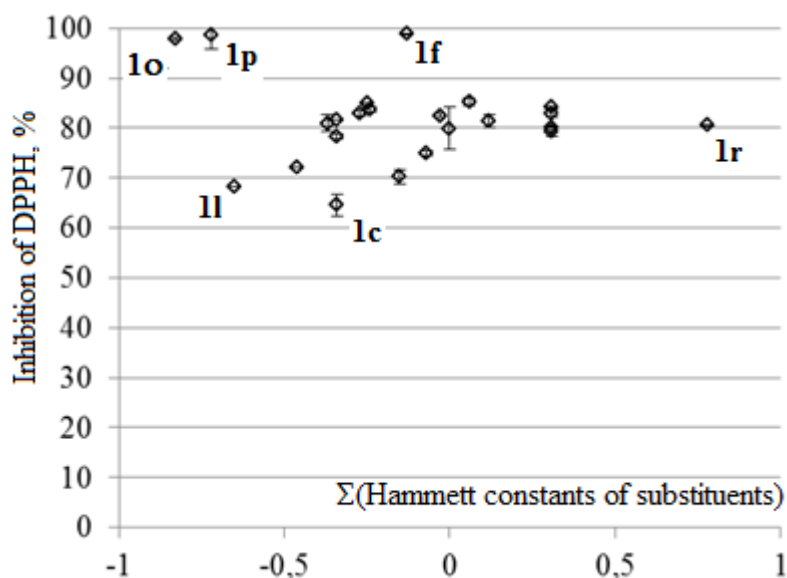


Figure S1. Correlation between electronic effects of the substituents and inhibition of DPPH, when molar ratio DPPH:antioxidant is 1:1.

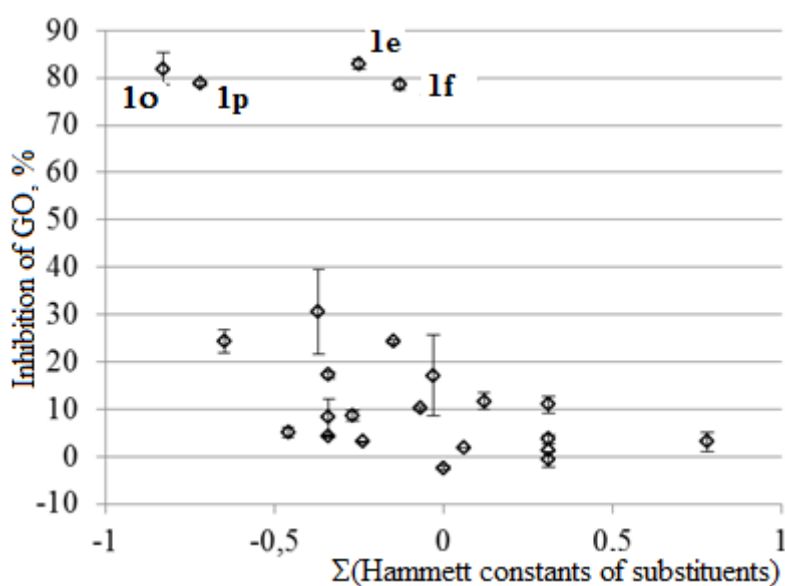


Figure S2. Correlation between electronic effects of the substituents and inhibition of GO, when molar ratio GO:antioxidant is 1:1.

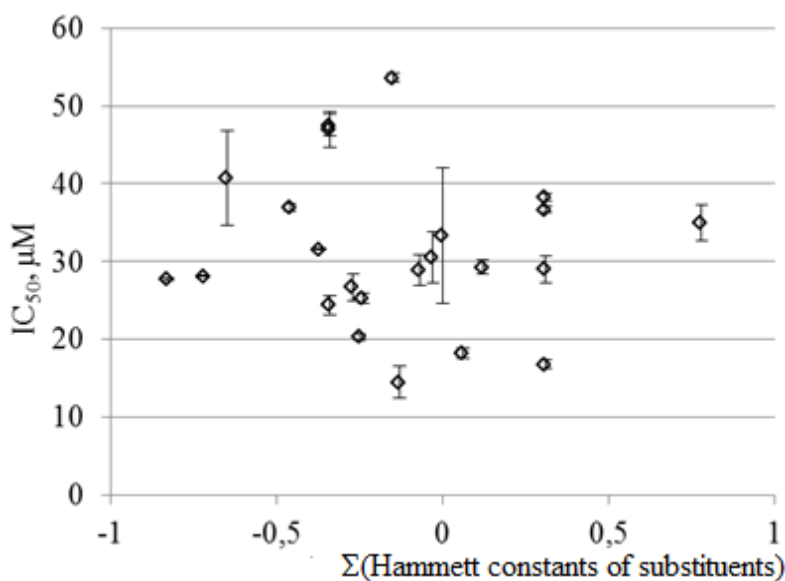


Figure S3. Correlation between electronic effects of the substituent and IC_{50} (DPPH assay)

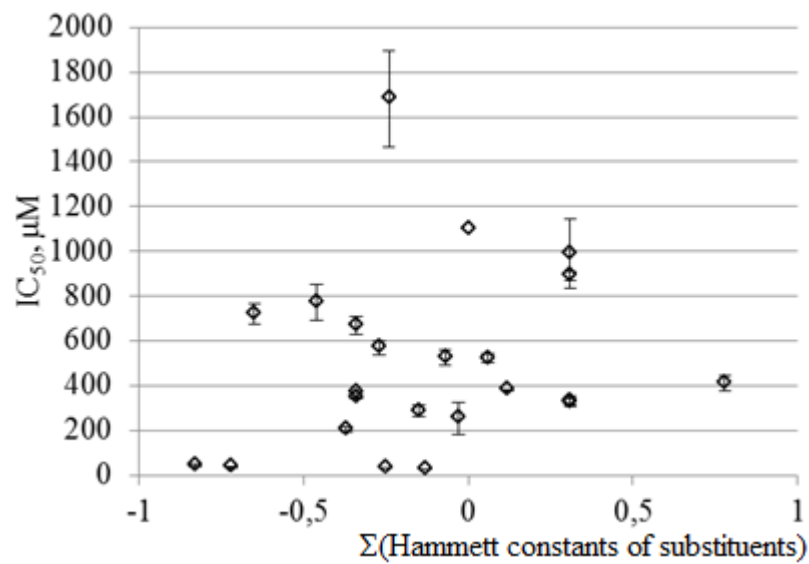


Figure S4. Correlation between electronic effects of the substituent and IC_{50} (GO assay)

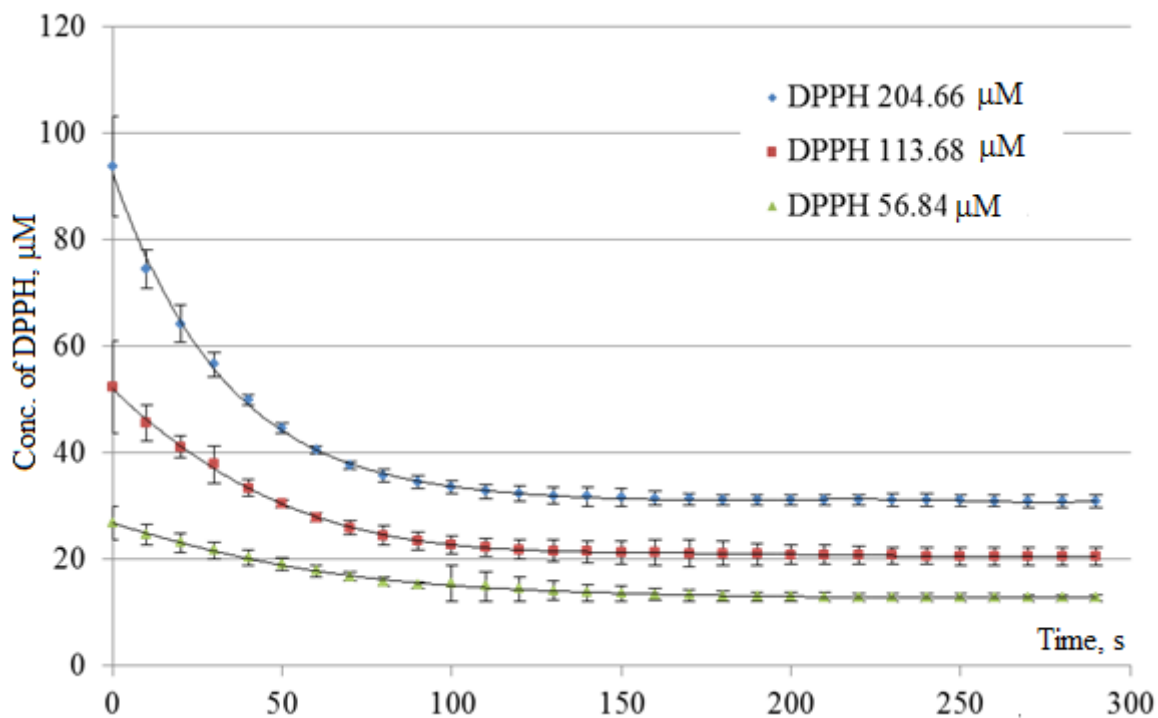


Figure S5. Kinetic curves of the reaction of arylmethyl Meldrum's acid **1 b** and DPPH

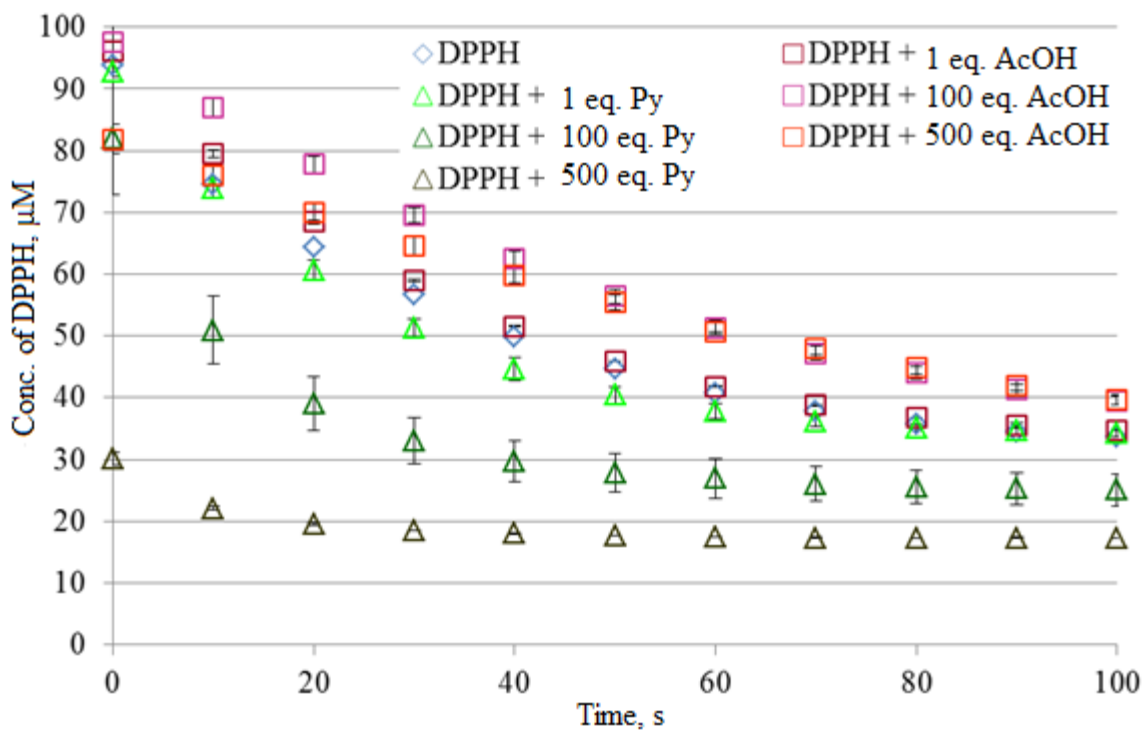


Figure S6. Kinetic curves of the reaction of arylmethyl Meldrum's acid **1 b** and DPPH with and without additives of pyridine and acetic acid (conc. of DPPH stock solution 204.66 μM)

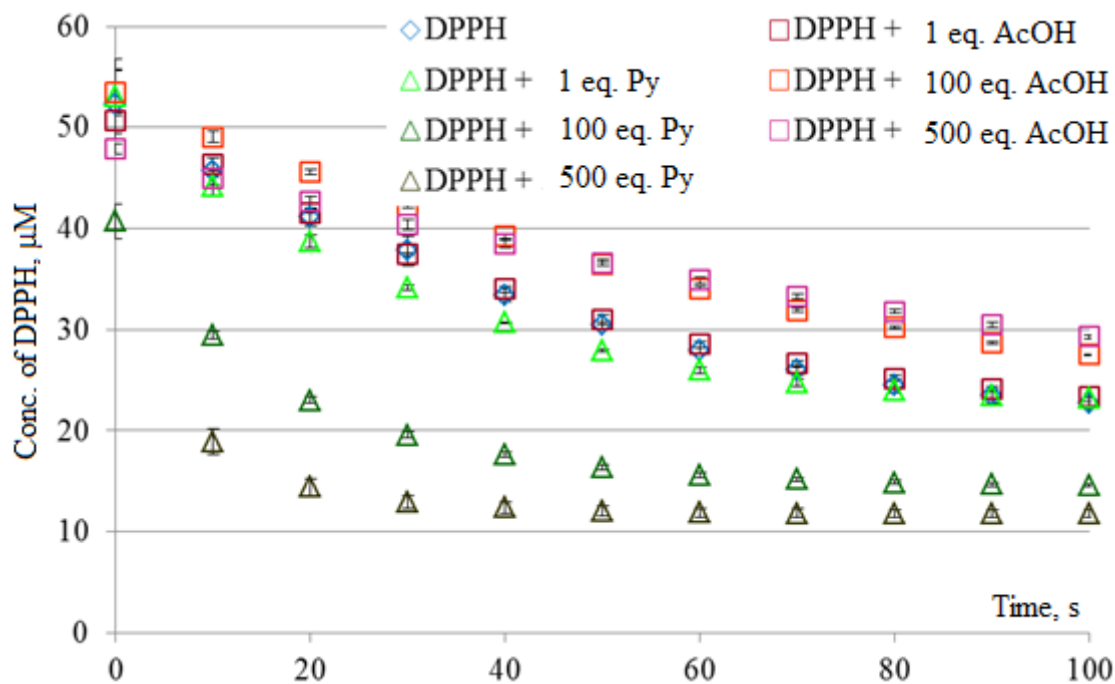


Figure S7. Kinetic curves of the reaction of arylmethyl Meldrum's acid **1 b** and DPPH with and without additives of pyridine and acetic acid (conc. of DPPH stock solution 113.68 μM)

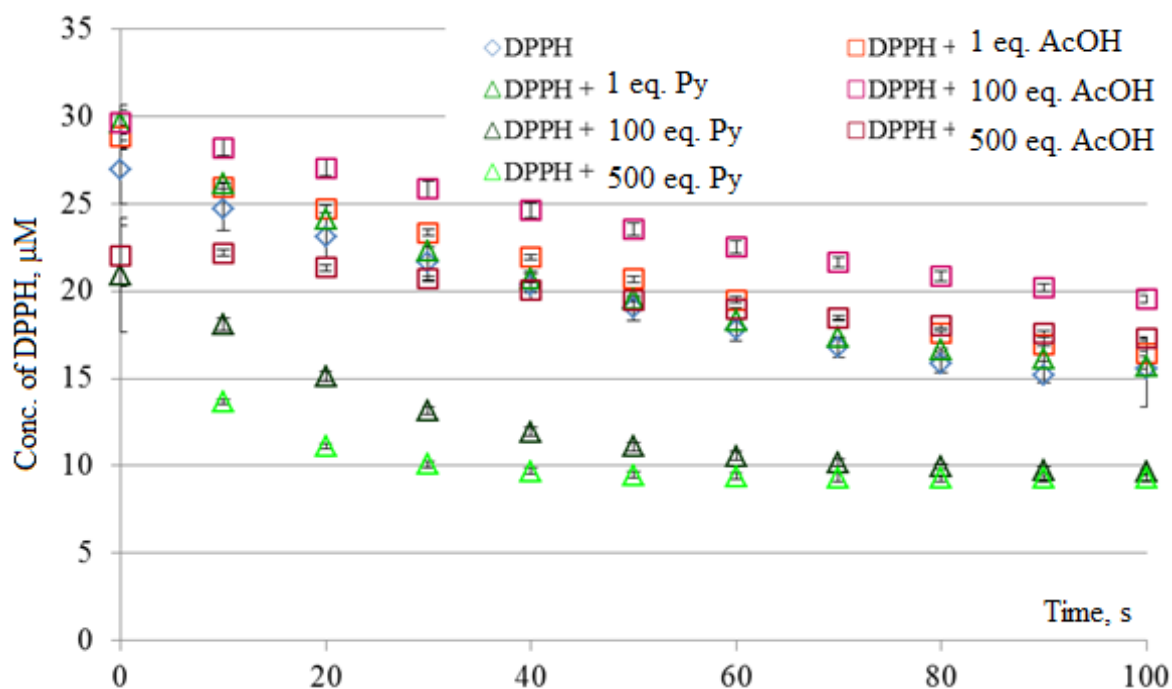


Figure S8. Kinetic curves of the reaction of arylmethyl Meldrum's acid **1 b** and DPPH with and without additives of pyridine and acetic acid (conc. of DPPH stock solution 56.84 μM)

Experimental procedure for the kinetic experiments

DPPH in ethanol (~ 56 , ~ 112 or ~ 205 μM ; 1.5 mL) and arylmethyl Meldrum's acid **1 b** (5850 μM ; 1.5 mL) with and without additive of pyridine (Py) or acetic acid (AcOH) (See Table S2) was mixed in cuvette and UV/Vis absorption was monitored every 10 s at 515 nm. The duration of experiment – 10 min.

Table S2

Entry	Solution of antioxidant	
	Additive	Amount, eq.
1	-	-
2	AcOH	1

3	AcOH	100
4	AcOH	500
5	Py	1
6	Py	100
7	Py	500

The amount of DPPH was calculated through calibration curve ($R^2 = 0.9991$):

$$c_{\text{DPPH}} = (A + 0.0642)/0.0114$$

Each experiment was carried triple. The results are presented as mean \pm standard deviation.

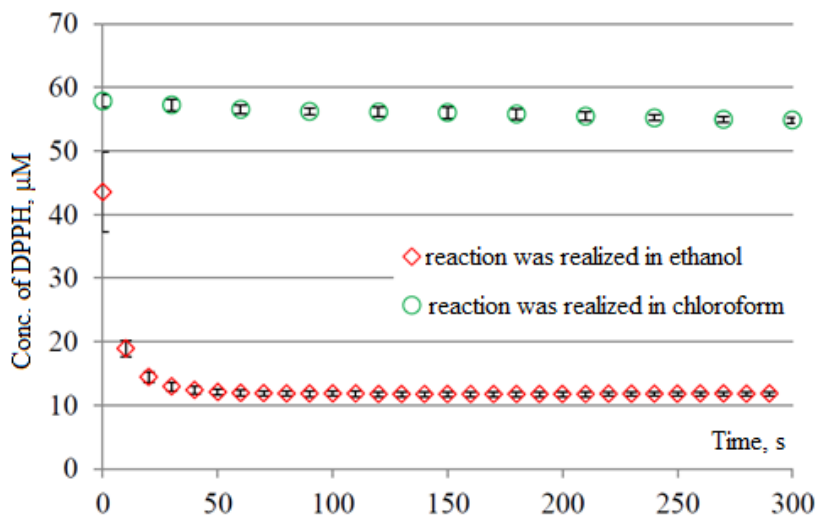


Figure S9. Kinetic curves of the reaction of arylmethyl Meldrum's acid **1 b** and DPPH in ethanol and chloroform

Experimental procedure

The experiments were realized analogous to the described above. When the reaction was realized in chloroform, both DPPH and arylmethyl Meldrum's acid was dissolved in chloroform. The duration of the experiment was 300 s.

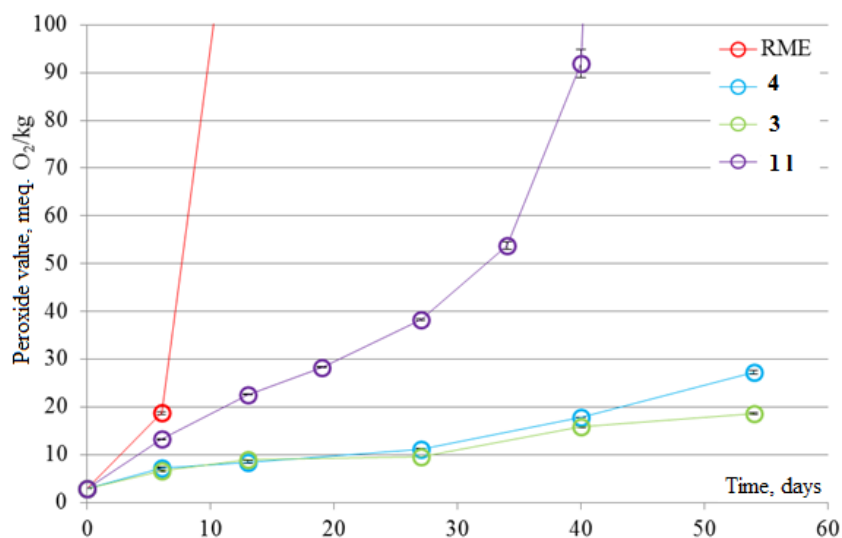


Figure S10. The impact of arylmethyl Meldrum's acids **11**, **3** and **4** on the oxidative stability of rapeseed oil methyl esters (RME).[†]

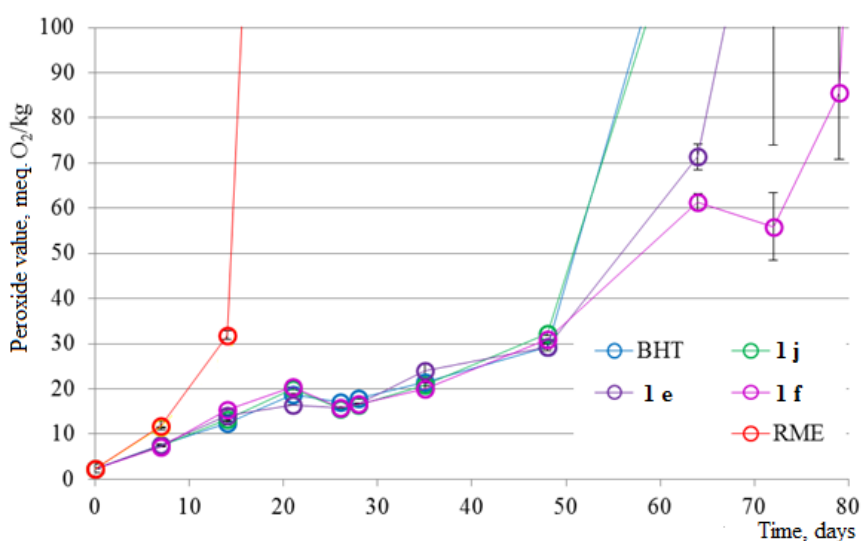


Figure S11. The impact of arylmethyl Meldrum's acids **1 e**, **f**, **j** and BHT on the oxidative stability of RME.

[†] The data of peroxide values are presented as mean \pm standard deviation from two unattached experiments

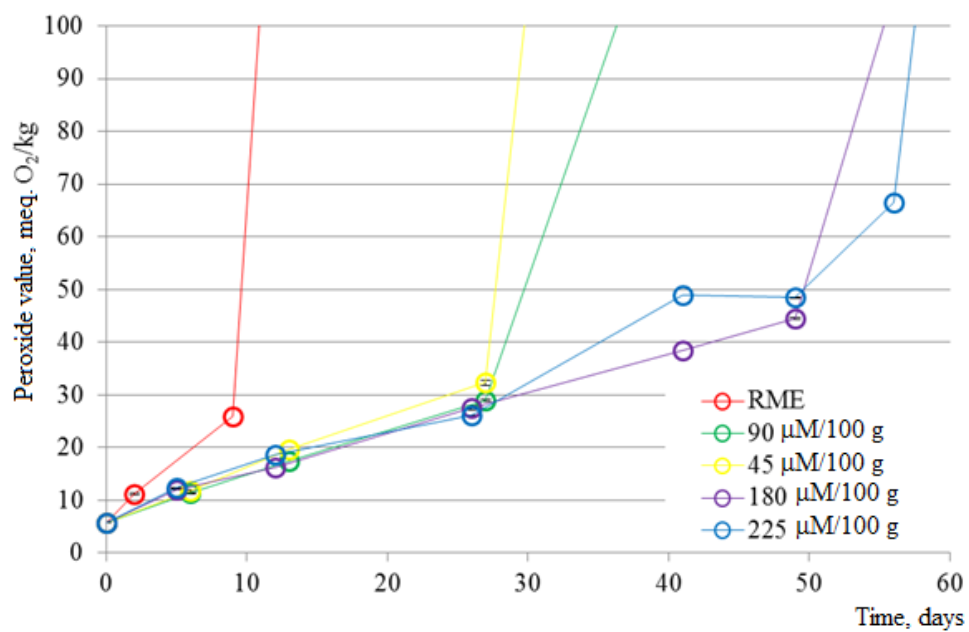


Figure S12. The impact of arylmethyl Meldrum's acids **1 j** (different concentrations) on the oxidative stability of RME.

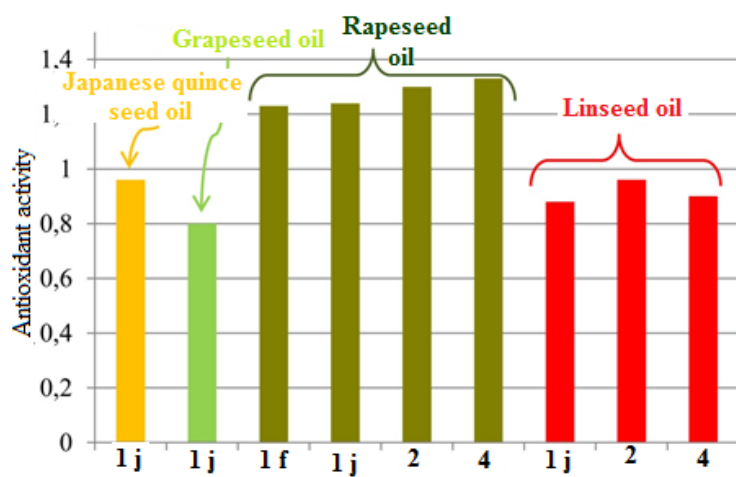
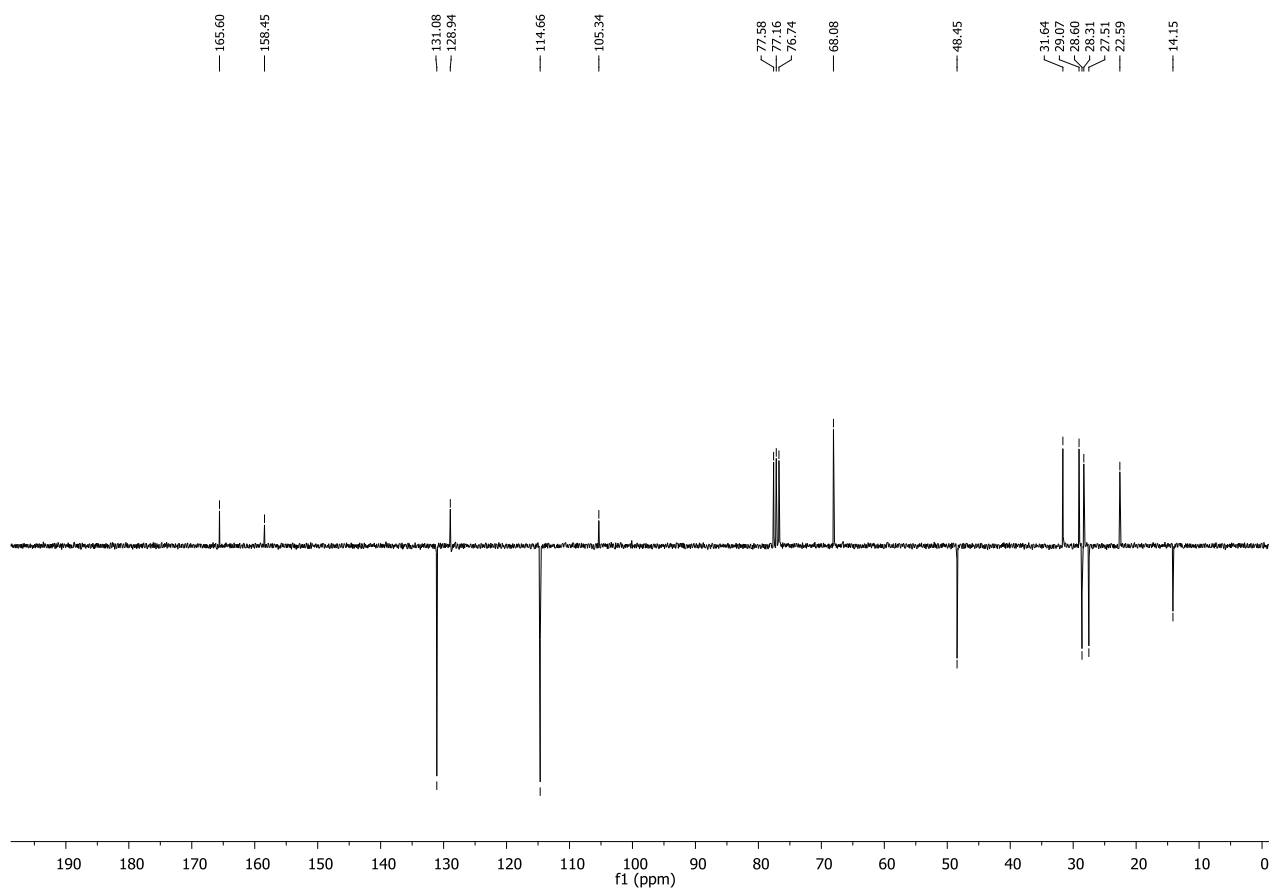
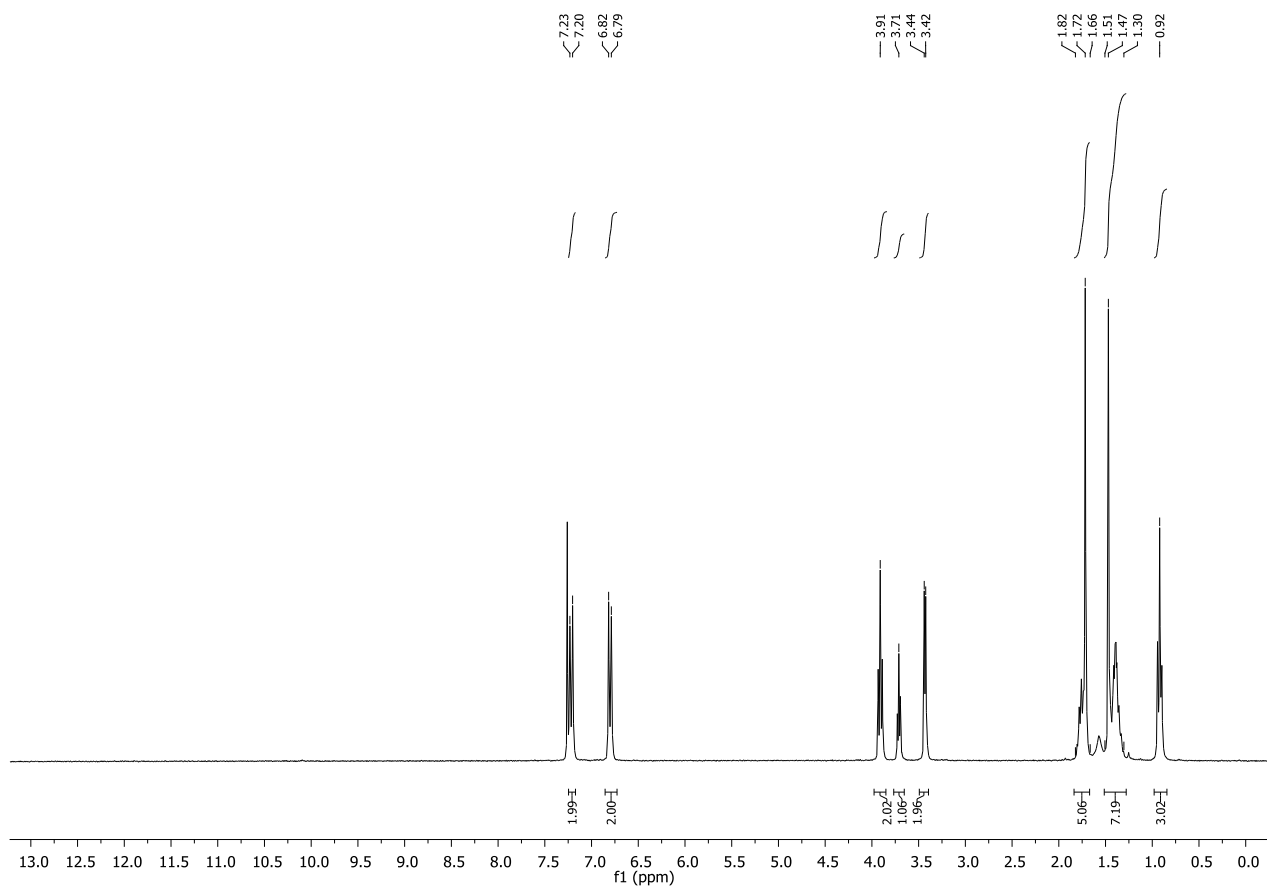


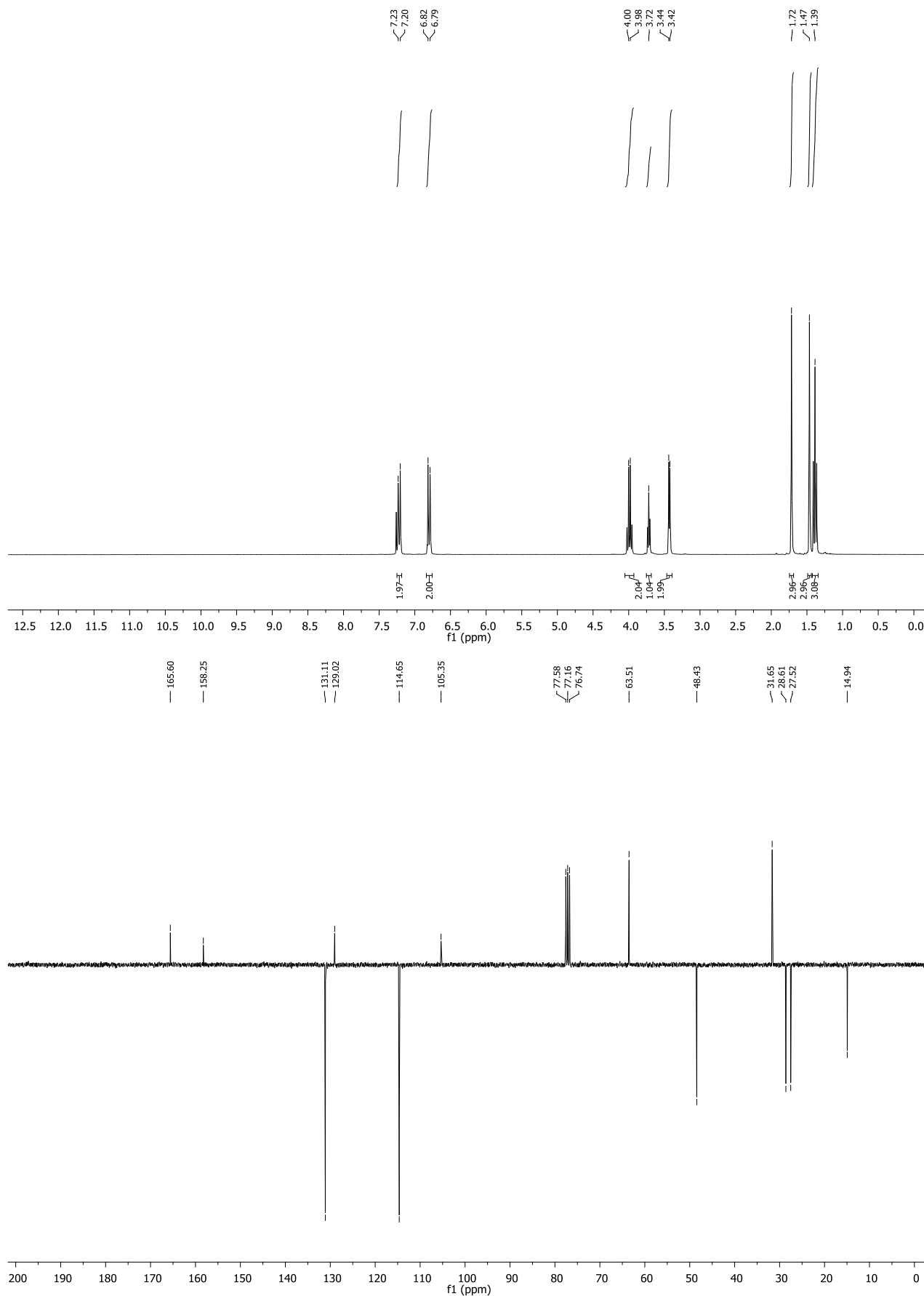
Figure S13. The impact of substituted Meldrum`s acids on the oxidative stability of different vegetable oils. Amount of the additive 90 $\mu\text{mol}/100\text{ g}$

NMR spectra of the new compounds

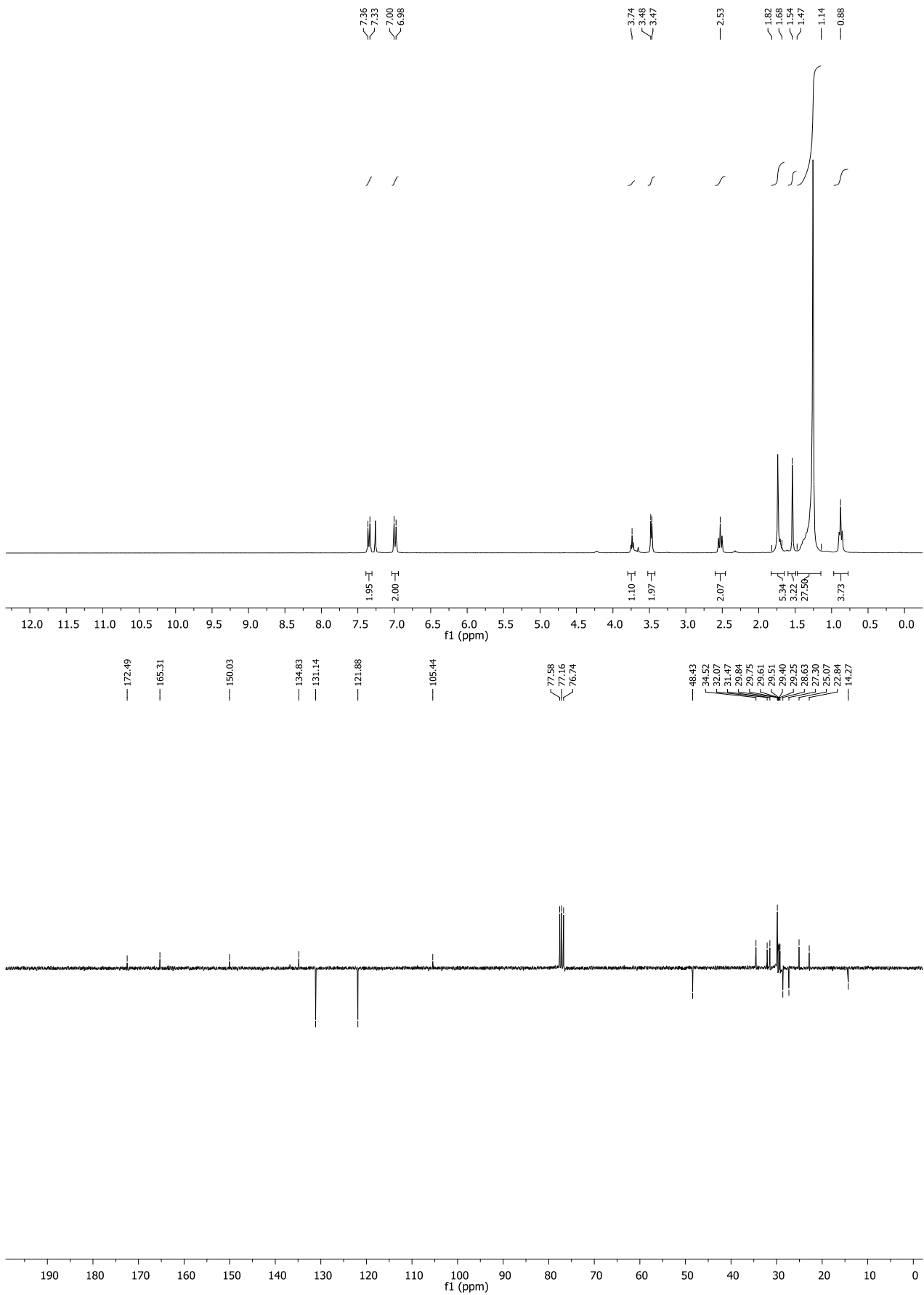
2,2-Dimethyl-5-(4-pentoxyphenyl)methyl-1,3-dioxane-4,6-dione (1j)



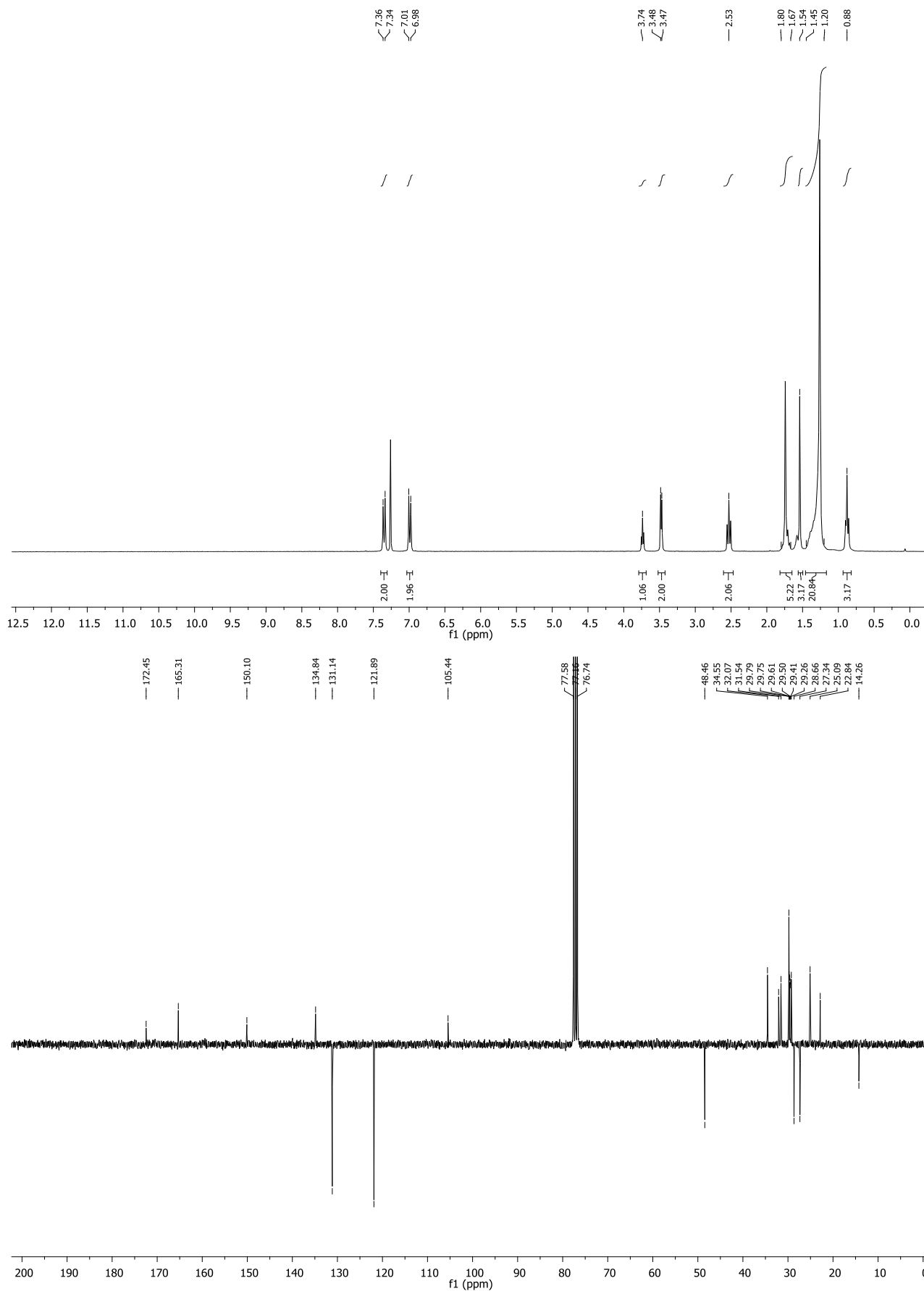
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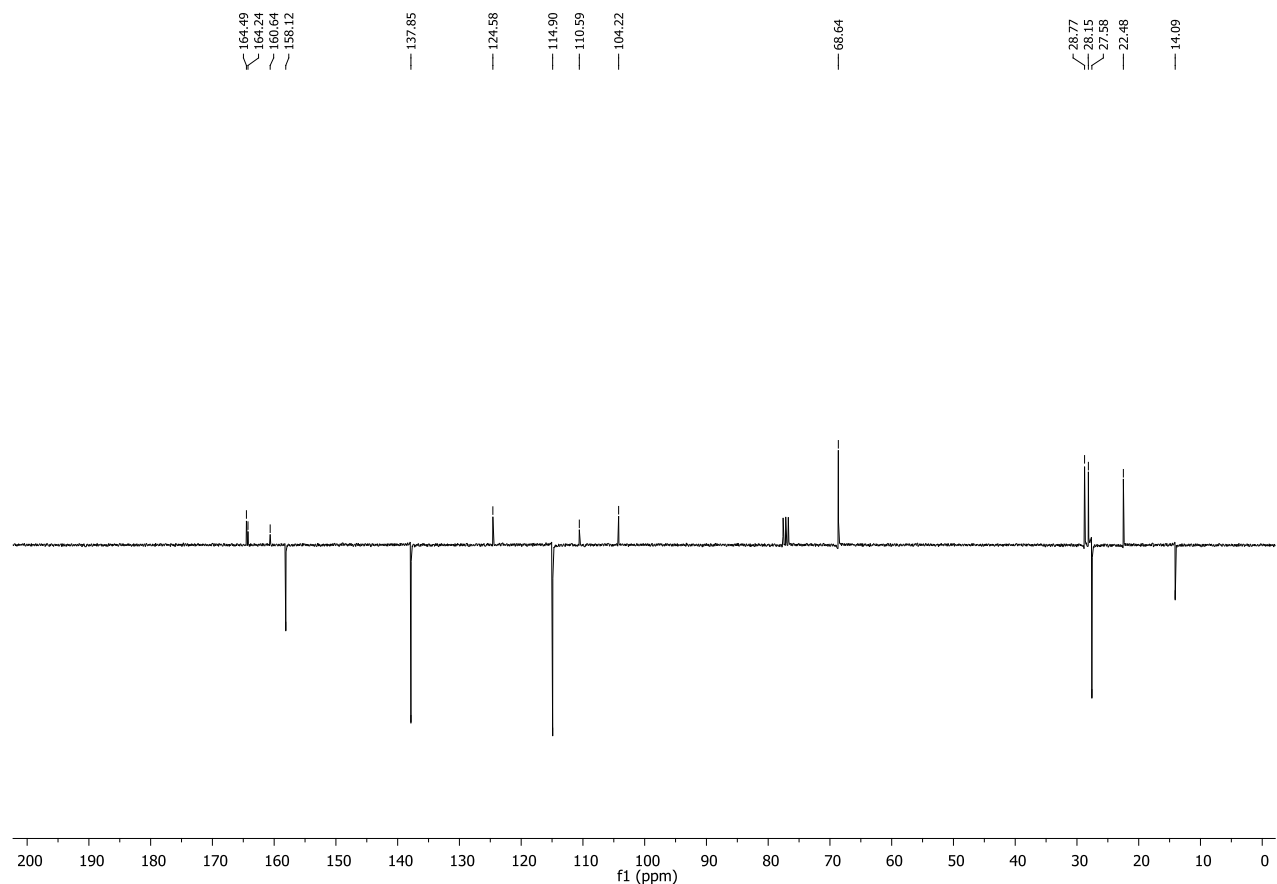
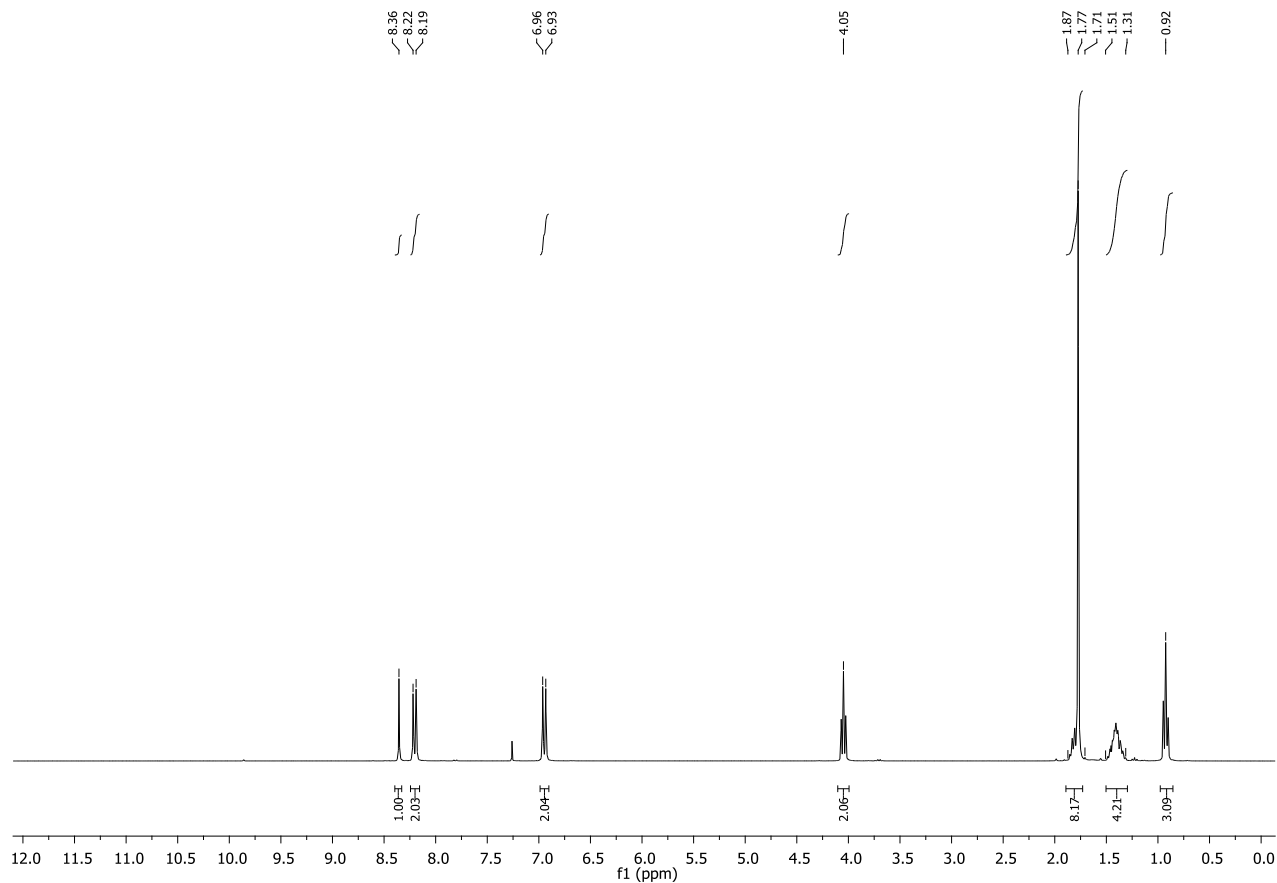
2,2-Dimethyl-5-(4-palmitoyloxyphenyl)methyl-1,3-dioxane-4,6-dione (1 u)



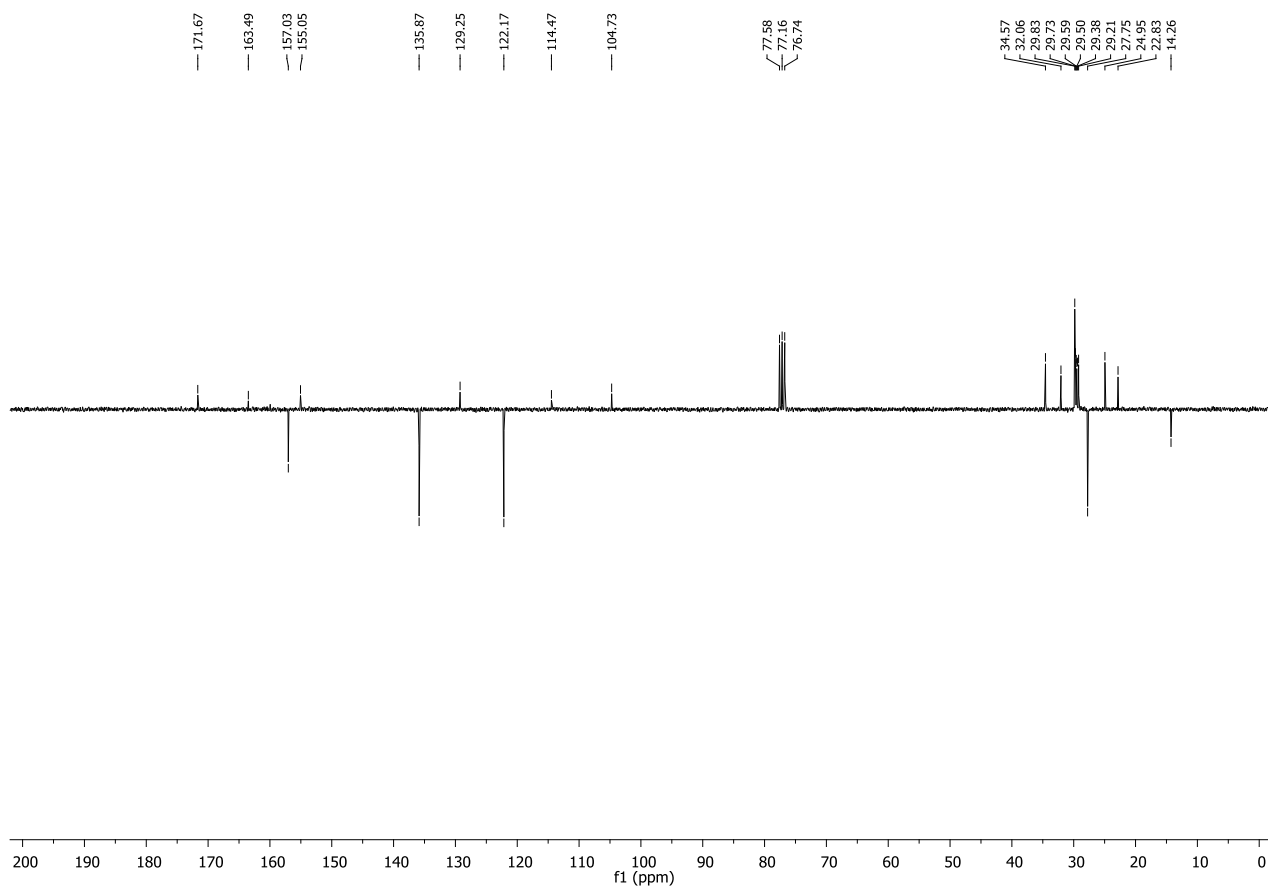
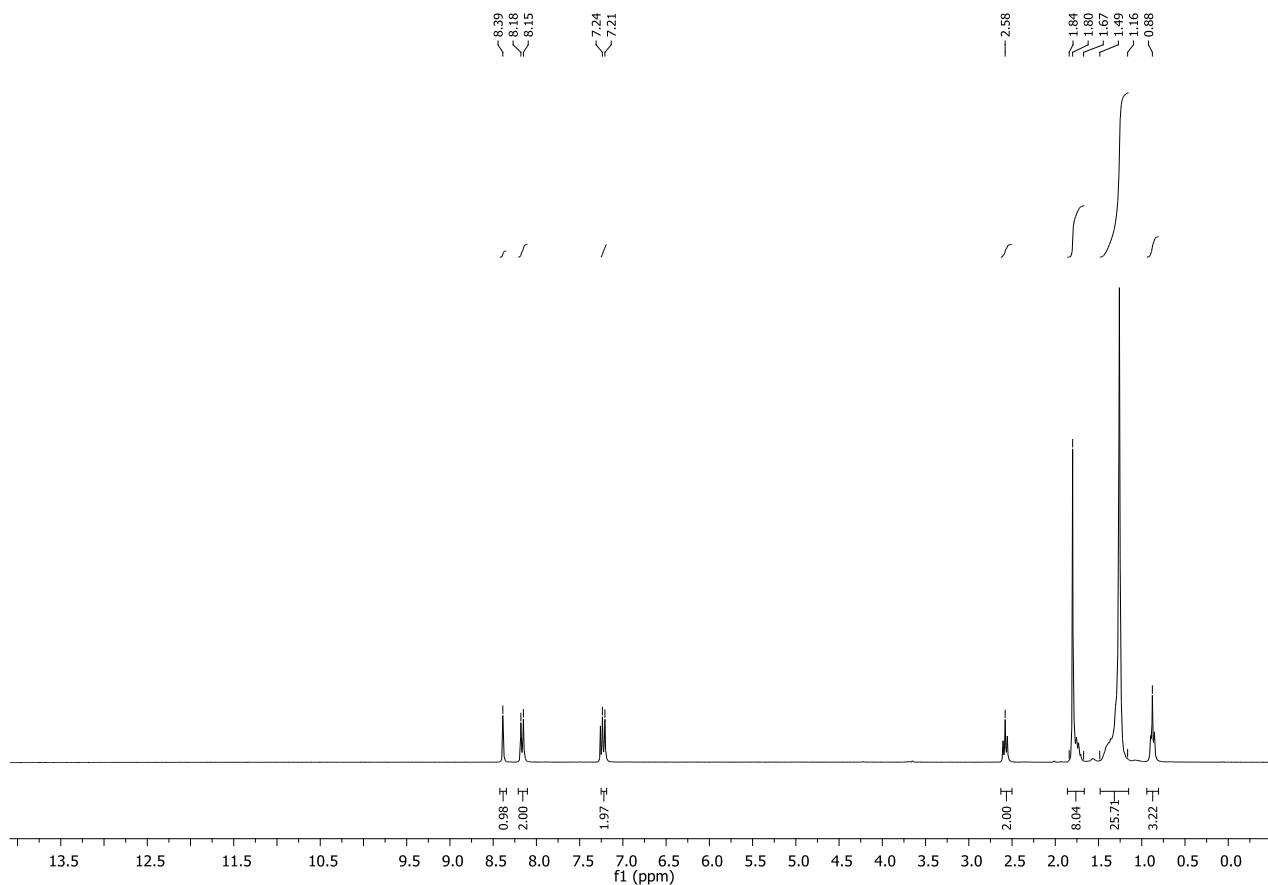
2,2-Dimethyl-5-(4-myristoyloxyphenyl)methyl-1,3-dioxane-4,6-dione (1 v)



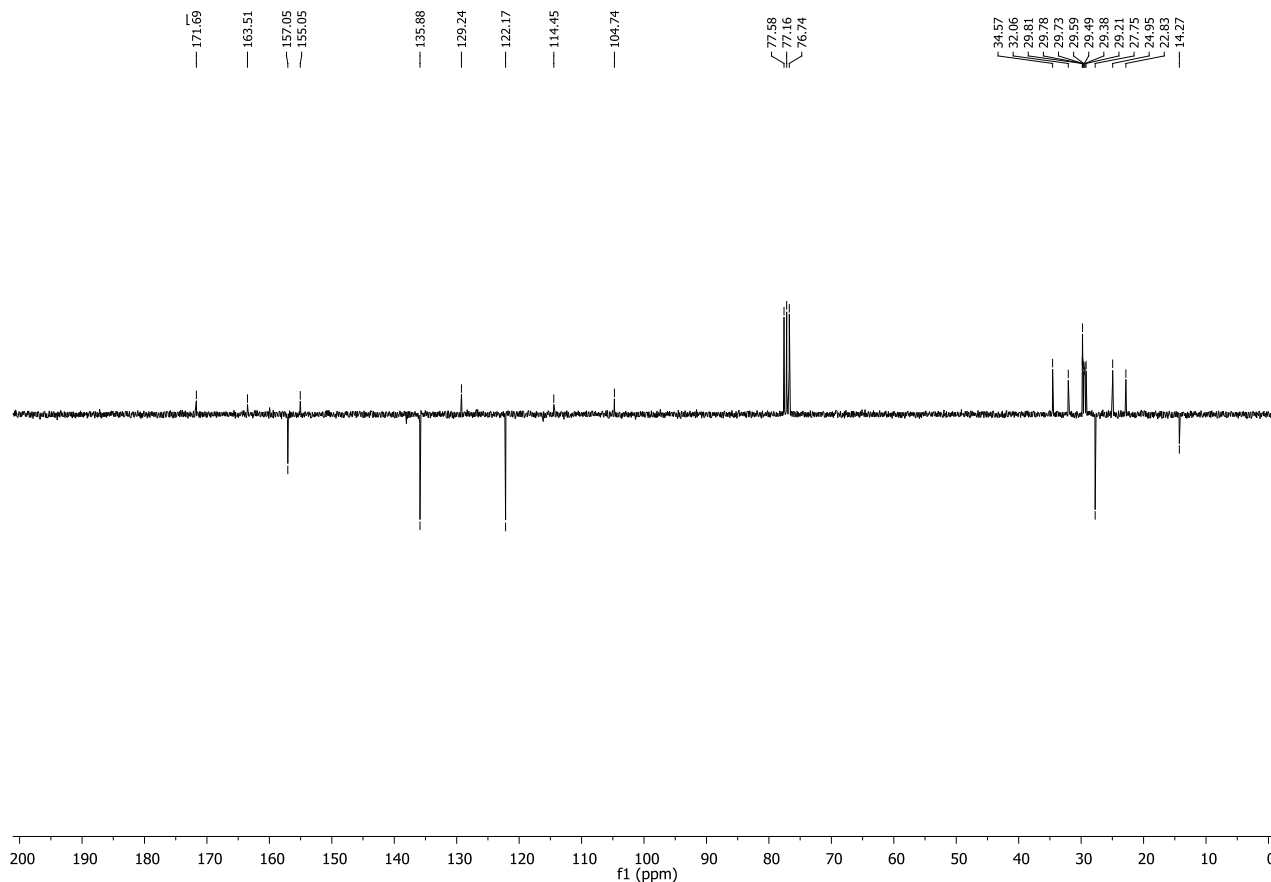
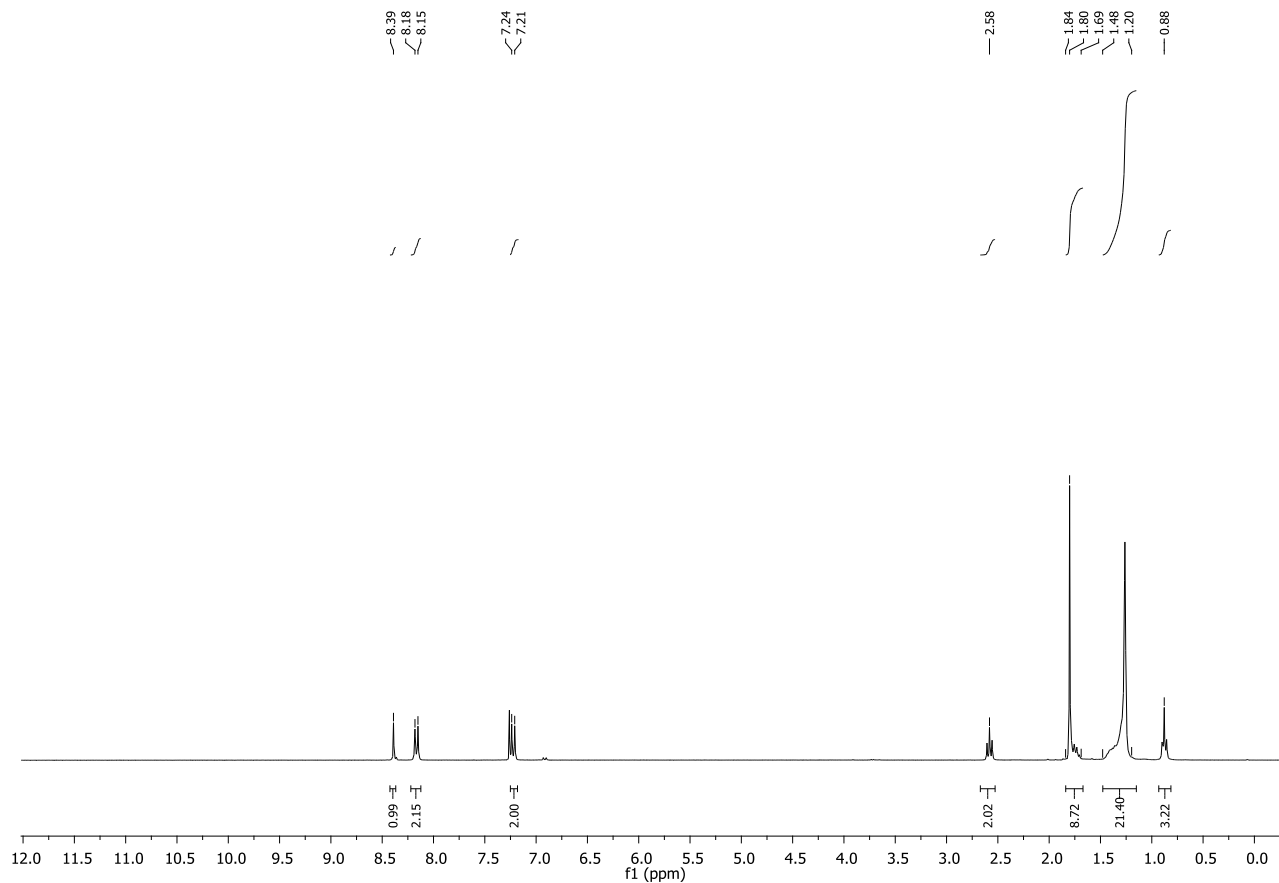
2,2-Dimethyl-5-(4-pentyloxyphenyl)methylene-1,3-dioxane-4,6-dione (10 j)



2,2-Dimethyl-5-(4-palmitoyloxyphenyl)methylene-1,3-dioxane-4,6-dione (10 u)



2,2-Dimethyl-5-(4-myristoyloxyphenyl)methylene-1,3-dioxane-4,6-dione (10 v)



5-(2,4-Dimethoxybenzyl)-2,2-dimethyl-5-[4-(1-phenyl-2-(2,4,6-trinitrophenyl)hydrazinyl)phenyl]-1,3-dioxane-4,6-dione (21)

