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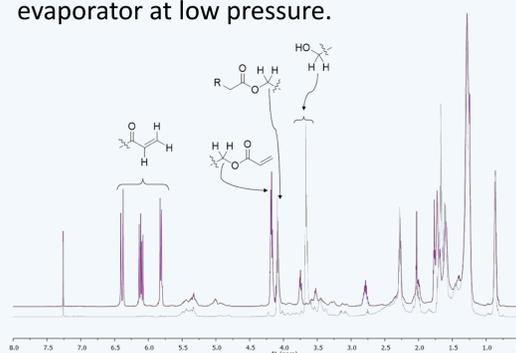
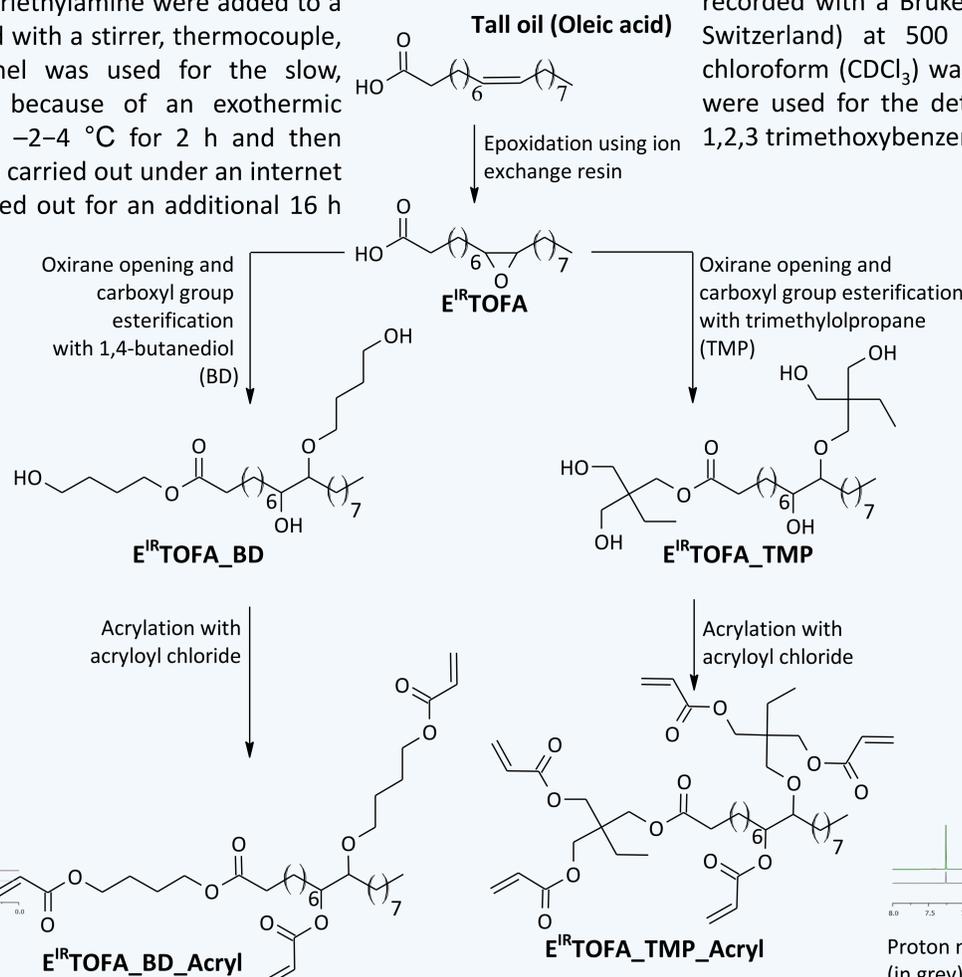
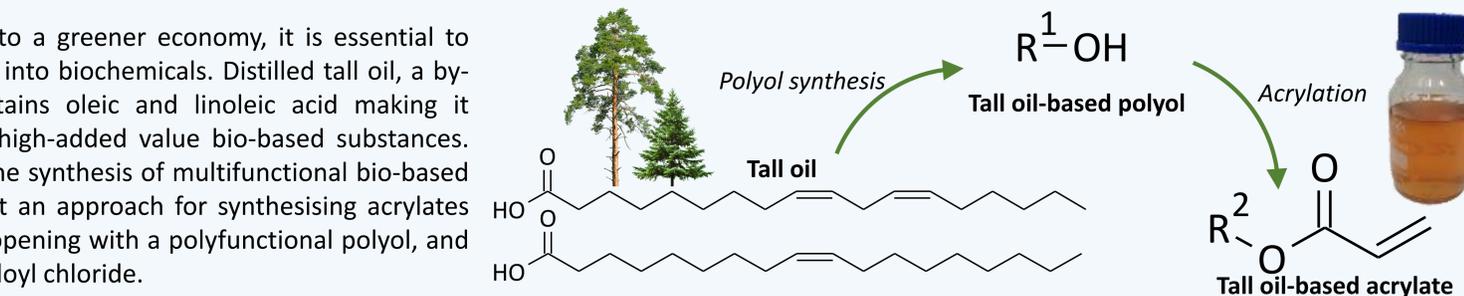
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To change the existing economic climate into a greener economy, it is essential to adapt the processing of forest bio-resources into biochemicals. Distilled tall oil, a by-product of the Kraft process, mainly contains oleic and linoleic acid making it attractive for the synthesis of a variety of high-added value bio-based substances. Recently, much attention has been paid to the synthesis of multifunctional bio-based acrylates for polymer production. We report an approach for synthesising acrylates from tall oil using epoxidation, oxirane ring-opening with a polyfunctional polyol, and following hydroxyl group acrylation with acryloyl chloride.

## Synthesis of tall oil-based acrylate

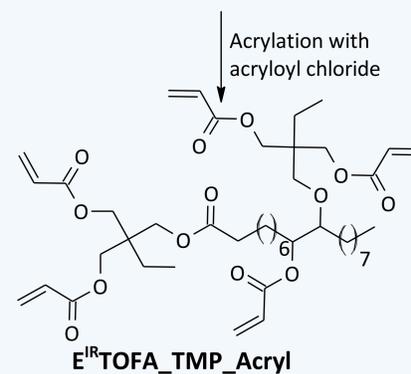
Previously synthesized polyol, E<sup>IR</sup>TOFA\_BD or E<sup>IR</sup>TOFA\_TMP, and a corresponding amount of ethyl acetate and triethylamine were added to a round bottom flask with four necks equipped with a stirrer, thermocouple, and a reflux condenser. A dropping funnel was used for the slow, intermittent pouring of acryloyl chloride because of an exothermic reaction. The temperature was maintained -2-4 °C for 2 h and then acryloyl chloride was added in. Synthesis was carried out under an internet atmosphere. Afterwards, synthesis was carried out for an additional 16 h at room temperature.

The molar ratio of polyol hydroxyl groups to acryloyl chloride was 1:1.05. The molar ratio of TEA to acryloyl chloride was 1:1. The mass ratio of polyol to ethyl acetate was 1:5. After synthesis, the mixture was filtrated and washed with a 0.5% sodium hydroxide solution and warm distilled water and dried with anhydrous sodium sulfate. The solvent was removed using a rotary evaporator at low pressure.



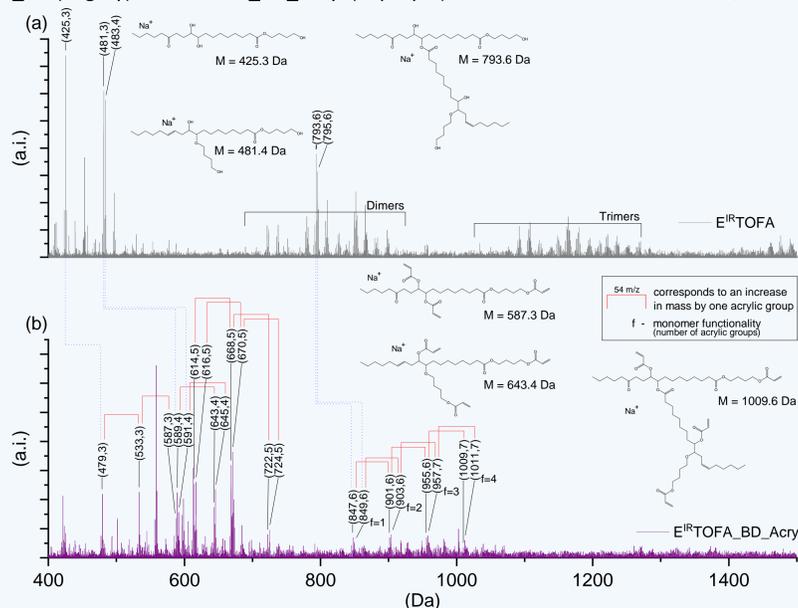
Proton nuclear magnetic resonance spectra for E<sup>IR</sup>TOFA\_BD (in grey) and E<sup>IR</sup>TOFA\_BD\_Acryl (in purple)

E<sup>IR</sup>TOFA\_BD\_Acryl

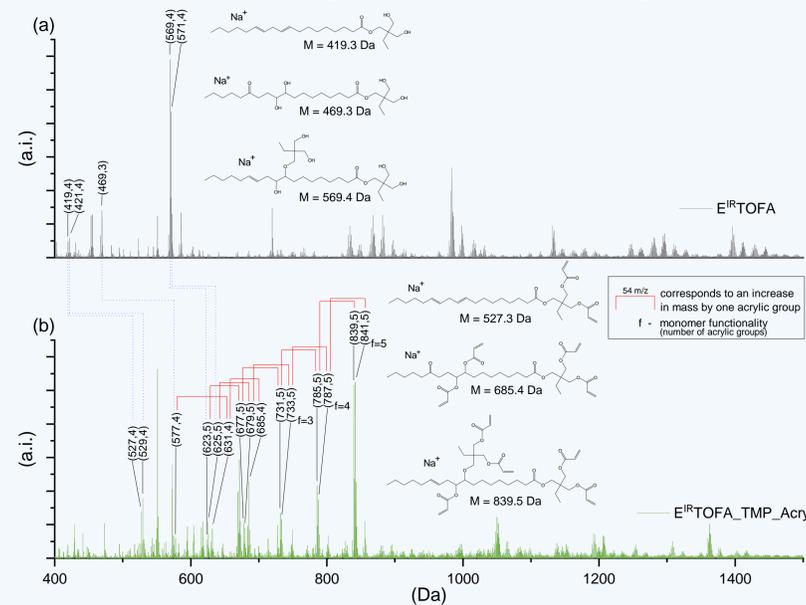


Proton nuclear magnetic resonance spectra for E<sup>IR</sup>TOFA\_TMP (in grey) and E<sup>IR</sup>TOFA\_TMP\_Acryl (in green)

E<sup>IR</sup>TOFA\_TMP\_Acryl



Matrix-assisted laser desorption/ionization-time of flight spectra of tall oil-based polyol and tall oil-based acrylates: a) E<sup>IR</sup>TOFA\_BD; b) E<sup>IR</sup>TOFA\_BD\_Acryl



Matrix-assisted laser desorption/ionization-time of flight spectra of tall oil-based polyol and tall oil-based acrylates: a) E<sup>IR</sup>TOFA\_TMP; b) E<sup>IR</sup>TOFA\_TMP\_Acryl

## Conclusions

Although synthesised E<sup>IR</sup>TOFA-based polyols consist of a mixture of various derivatives, they contain free hydroxyl groups, which can be converted to acrylic groups. Considering the proton nuclear magnetic resonance data and Matrix-assisted laser desorption/ionization-time of flight spectra, the synthesised acrylates from tall oil fatty acids-based polyols contain several acrylic groups required for polymerisation according to the Michael reaction mechanism. E<sup>IR</sup>TOFA\_BD\_Acryl contains 0.0039 mol/g acrylic groups, but E<sup>IR</sup>TOFA\_TMP\_Acryl contains 0.0035 mol/g acrylic groups. The obtained tall oil polyol-based acrylates are suitable for polymer synthesis by the Michael reaction. Moreover, the synthesised tall oil polyol-based acrylates are promising in producing other polymers, for example, in UV-initiated free radical polymerisation in the production of coatings.

## Acknowledgements

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