

# 10<sup>TH</sup> PAUL WALDEN SYMPOSIUM ON ORGANIC CHEMISTRY

**Programm and Abstracts** 

Riga June 15-16, 2017

#### Friday, June 16 Professor Olafs Daugulis (University of Houston, Houston, USA) 9.00-10.00 New methods for carbon-hydrogen bond functionalization 10.00-10.30 Dr. Vilnis Liepiņš (JSC Olainfarm, Olaine, Latvia) The use of DoE in chemical process optimization Dr. Ronalds Zemribo (Latvian Institute of Organic Synthesis, Riga, Latvia) 10.30-11.00 Discovery of Remeglurant, a potent and selective mGluR5 receptor negative allosteric modulator 11.00-11.30 Coffee break 11.30-12.30 Poster Session II 12.30-13.45 Lunch break 13.45-14.45 Professor Roger Strömberg (Karolinska Institute, Department of Biosciences and Nutrition, Huddinge, Sweden) Modification and conjugation of oligonucleotides for development of oligonucleotide therapeutics 14.45-15.10 Coffee break Doctoral student Jevgeņija Lugiņina (RTU) 15.10-15.30 Aziridine ring opening and other reactions in liquid sulfur dioxide 15.30-15.50 Doctoral student Mārtiņš Otikovs (Latvian Institute of Organic Synthesis, Riga, Latvia) Towards spinning of artificial spider silk Dr. Helmärs Šmits (Syngenta Crop Protection, Basel, Switzerland) 15.50-16.30 Agrochemical process research: synthesis of aryl 1,3-diones and stereoselective cyclopropanation

16.30-16.45 Closing remarks, Award ceremony for poster prizes

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## **Aziridine Ring Opening and Other Reactions** in Liquid Sulfur Dioxide

Jevgeņija Lugiņina

Faculty of Materials Science and Applied Chemistry, Riga Technical University, Paula Valdena Str. 3, Riga, LV-1048, Latvia jevgenija.luginina@rtu.lv

In recent years, many applications of SO2 and its surrogates in organic synthesis have been reported [1]. Due to high polarity and Lewis acid properties sulfur dioxide can be used as strongly ionizing solvent. Furthermore, it has a high dipole moment (1.61 D), therefore it readily dissolves both organic and inorganic compounds. On the other hand, SO2 has been reported as a reaction medium for processes involving carbenium ions. Surprisingly, despite all these facts sulfur dioxide has not entered the "classic toolbox" of synthetic organic chemist as an everyday-solvent. This has prompted us to search for organic reactions that would profit from their running in liquid SO2 as a reaction medium.

We have discovered that different aziridines and azetidines undergo efficient ring-opening reactions in liquid SO2 with metal halides and thiols as nuchleophiles (Scheme 1) [2,3]. Transformations proceed stereoselectively without racemization of chiral center in the case of enantiomerically pure protected or unprotected heterocycles.

Scheme 1. Nucleophilic ring opening of small heterocycles in liquid SO2.

We investigated glycosylation reaction with a wide range of O-, and S-nucleophiles of different monosaccharides in liquid SO2 (Scheme 2). Employing SO2 as a reaction media and glycosyl fluorides (LG = F) as glycosyl donors we performed nucleophilic displacement without any additional promoter or Lewis acid additive.

$$(PGO)_4 \xrightarrow{N_1}_{LG} \frac{NuH}{SO_2} (PGO)_4 \xrightarrow{NU}_{NU} \frac{O}{NU}$$

Scheme 2. Formation of glycosyl bond in liquid SO2.

A novel method for the synthesis of sulfones also has been elaborated. The first step in this sequence is bora-ene reaction of sulfur dioxide and substituted potassium trifluoroborate giving a sulfinate moiety complexed to borate center (Scheme 3). It was observed that crotyl and prenyl systems react at their γ-position with SO<sub>2</sub>. This serves as evidence for the ene-type regioselectivity. In the second step the alkylation of sulfinate moiety provides expected sulfones.

$$R^1$$
 $R^2$ 
 $BF_3K$ 
 $SO_2$ 
 $R^1$ 
 $R^2$ 
 $OBF_3K$ 
 $R^3$ 
 $R^3$ 

Scheme 3. Sulfone synthesis using bora-ene reaction as a key step.

Supervisor: Dr. chem. M. Turks

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